Long-term measurements of glyoxal (CHOCHO) and formaldehyde (HCHO) M. Vrekoussis, F. Wittrock, A. Richter, A.Ladstätter-Weißenmayer and J.P. Burrows Institute of Environmental Physics and Remote Sensing, University of Bremen, P. O. Box 330440, D-28334 Bremen, Germany



ABSTRACT

It is well known that **VOC** are released into the atmosphere by both anthropogenic and natural sources (e.g. fossil fuel combustion, biomass burning and biogenic emissions). Nevertheless the discrimination of these sources is difficult due to the overall great uncertainty of their emissions. In order to better characterize these sources, tracers of the oxidation of the VOCs such as HCHO and CHOCHO are used as proxies. Formaldehyde, the smallest aldehyde in the atmosphere is primarily formed through the oxidation of VOC by OH but is also directly emitted. Glyoxal, the smallest dicarbonyl, is a mutagenic product formed mainly under the chemical reactions of VOC with OH radicals. This work presents for the first time concurrent results of both species measured on a global scale for a period of 4 years covering the period of 2003-2006. The results were obtained by applying the differential optical absorption technique (DOAS) to spectra measured by the satellite instrument SCIAMACHY. HCHO has been analyzed in the spectral region of 334.3-348.5 nm to avoid any correlation with an instrument grating polarisation structure around 360 nm. CHOCHO was retrieved in the blue spectral range at **436.0–457.0 nm**. The annual and seasonal variation of these species is examined above some photochemical hot spots induced by anthropogenic and biogenic activities. Some of these areas of interest are the Central Africa, South America (mainly Brazil), India, China and Indonesia as well as some major cities of Europe and the United States. During the period 2003-2006 and on an annual basis the vertical column (VC) values of formaldehyde ranged from **DL (4.10¹⁵) – 2.5[.]10¹⁶ molecules.cm⁻²** while the respective glyoxal values ranged from **DL (2.10¹⁴) – 2.0.10¹⁵ molecules.cm⁻²**.



RESULTS - ANNUAL VARIATION

Glyoxal and formaldehyde were calculated on an annual basis for the period **1.8.2002** – **31.12.2006**. The figures to the left depicts the global multi-annual composite for the two species. Certain areas appear to have concurrent enhanced vertical column values for both CHOCHO and HCHO pointing to the presence of photochemical hot spots. Some of these areas are found at **South** America, Africa, East USA, the developing Asian cities, India and to a lesser extent Europe. At a first glance it was found that the highest glyoxal values were observed near the various source regions (anthropogenic, biogenic and biomass **burning**). This could be associated with the short lifetime of the CHOCHO.

The graph to the right presents the annual variation of glyoxal above four indicative areas where glyoxal a) had increased during the last years (Tianjin), b) maintained its levels (Kolkata), did not have a seasonal variation (**Bangkok**) and d) had winter maxima (Southern Hemisphere, Johannesburg).





respectively.















The main known **sinks** of HCHO and CHOCHO are: a) the reaction with the **OH** radicals and the b) photolysis leading to an estimated lifetime of 3.5h and 1.5h



SCIAMACHY (Scanning Imaging Absorption Spectrometer for Atmospheric CHartographY) is an imaging spectrometer whose primary mission objective are global measurements of trace gases in the troposphere and in the stratosphere. The solar radiation transmitted, backscattered and reflected from the atmosphere is recorded at relatively high resolution (0.2 nm to 1.5 nm) over the range 240 nm to 1700 nm, and in selected regions between 2.0 µm and 2.4 µm. Sciamachy has a global coverage of 6 days with a spatial resolution of 60kmx30km.

INSTRUMENTATION & EXPERIMENT



The vertical columns (VC) of HCHO and glyoxal are calculated with the Differential Optical Absorption Spectroscopy (DOAS) and subsequently applying the air mass factor correction (AMF, calculated by the radiative transfer model SCIATRAN) to the slant columns (SC). The latter is the integrated amount of absorber averaged over all light paths. HCHO has analyzed in the spectral region of 334.3–348.5 nm to avoid any correlation with an instrument grating polarisation structure around 360 nm. CHOCHO was retrieved at the blue spectral range at 436.0–457.0 nm. The main steps of the retrieval technique are listed below:

- The logarithms of the earthshine spectra and the solar irradiance are normalised to produce an absorption spectrum.
- Broadband features due to Mie and Rayleigh scattering are removed via the fitting of a low order polynomial.
- Ring spectrum is fitted to correct the rotational and vibrational Raman Scattering
- Other absorbers (in addition to the species of interest) interfering at the above mentioned wavelengths are fitted. In case of HCHO and CHOCHO these were the BrO, O₃, NO₂, H₂O and O₄. The SC are calculated by applying the Lambert-Beer law
- $\overrightarrow{}$ The final step involves the conversion of the SC to VC via the A

The retrieved SCIAMACHY data of CHOCHO and HCHO, for the period 2003 – 2006 were also analyzed on a seasonal basis by calculating the average value of winter months (blue color, $\gamma^{-8.0 \times 10}$ left figure) and summer months (**red color**, left figure). It was found that in most of the cases, both glyoxal and formaldehyde show similar seasonal behavior. For instance, during winter, the observed maxima are found in the southern hemisphere, while during summer the northern hemisphere dominates. This enables us to draw the conclusion that both species have short as well comparable lifetimes.

For the areas chosen (see the analysis below), the mean monthly value of CHOCHO and HCHO was calculated (right graph). It can be seen that during warm conditions both species maximize. This is mainly due to the **enhanced** OH that controls the VOC oxidation and produces the HCHO and the CHOCHO.



VC-CHOCHO-SH: 3.14·10¹⁴ 8.71·10¹ VC-HCHO-NH: 8.63.10¹⁵ 1.26.10¹⁵ VC-HCHO-SH: 1.01.10¹⁶ 1.95.10¹⁵ All values are in molecules cm⁻²