

Global tropospheric NO₂ column distributions: Comparing three-dimensional model calculations with GOME measurements

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Abstract. Tropospheric NO₂ columns derived from the data products of the Global Ozone Monitoring Experiment (GOME), deployed on the ESA ERS-2 satellite, have been compared with model calculations from two global three-dimensional chemistry transport models, IMAGES and MOZART. The main objectives of the study are an analysis of the tropospheric NO₂ data derived from satellite measurements, an interpretation of it and evaluation of its quality using global models, and an estimation the role of NO₂ in radiative forcing. The measured and modeled NO₂ columns show similar spatial and seasonal patterns, with large tropospheric column amounts over industrialized areas and small column amounts over remote areas. The comparison of the absolute values of the measured and modeled tropospheric column amounts are particularly dependent upon uncertainties in the derivation of the tropospheric NO₂ columns from GOME and the difficulty of modeling the boundary layer in global models, both of which are discussed below. The measured tropospheric column amounts derived from GOME data are of the same order as those calculated by the MOZART model over the industrialized areas of the United States and Europe, but a factor of 2-3 larger for Asia. The modeled tropospheric NO₂ columns from MOZART as well as the column amounts measured by GOME are in good agreement with NO₂ columns derived from observed NO₂ mixing ratios in the boundary layer in eastern North America. The comparison of the models to the GOME data illustrates the degree to which present models reproduce the hot spots seen in the GOME data. The radiative forcing of NO₂ has been estimated from the calculated tropospheric NO₂ columns. The local maxima in the radiative forcing of tropospheric NO₂ for cloud-free conditions over the eastern United States and western Europe represent 0.1-0.15 W m⁻², while values of 0.04-0.1 W m⁻² are estimated on a continental scale in these regions, of the same order of magnitude as the forcing of N₂O and somewhat smaller than the regional forcing of tropospheric ozone. The globally averaged radiative forcing of tropospheric NO₂ is negligible, ~0.005 W m⁻².

1. Introduction

Nitrogen dioxide (NO₂) plays a key role in tropospheric chemistry. For example, its photolysis leads to the formation of ozone (O₃) and its reaction with the hydroxyl radical (OH) produces nitric acid (HNO₃). These processes are of importance in both the planetary boundary layer and free

troposphere [e.g., Chameides *et al.*, 1992]. Thereby NO₂ participates in the control of the strong oxidant, O₃, and the strongest atmospheric oxidizing agent, OH, and plays an important part in determining the oxidizing capacity of the atmosphere [Logan *et al.*, 1981]. Recently, it has been recognized that NO₂ contributes both directly and indirectly to the radiative forcing of climate [Solomon *et al.*, 1999]. In addition, it is known that in high concentration, NO₂ causes respiratory problems for humans [e.g., *Environmental Protection Agency (EPA)*, 1998].

As both the atmospheric sources and sinks of NO₂ are strongly non homogeneously distributed, the concentration of NO₂ in the troposphere is highly variable, both spatially and temporally [e.g., Emmons *et al.*, 1997]. In the boundary layer above industrialized areas, values up to several tens of ppbv are observed having a strong diurnal variation, whereas in remote areas the concentrations are typically in the range of 10-200 pptv. Prior to the advent of space-based remote sensing instrumentation such as the Global Ozone Monitoring Experiment (GOME), and because of its high intrinsic variability and consequent sparse set of in situ or remote sensing measurements, global knowledge about the amount

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and distribution of NO₂ has been difficult if not impossible to obtain. In several recent studies [Horowitz and Jacob, 1999; Levy et al., 1999; Penner et al., 1998; Wang et al., 1998; Kraus et al., 1996; Lamarque et al., 1996], chemistry transport models have been used to calculate the global three-dimensional distribution of tropospheric NO_x and study its sources, sinks, and atmospheric transport.

GOME [Burrows et al., 1991, 1993, 1999] is a small scale version of the Scanning Imaging Absorption Spectrometer for Atmospheric Chartography (SCIAMACHY) [Bovesmann et al., 1999] observing between 232 and 793 nm and only in nadir viewing geometry. It measures the radiance upwelling from the atmosphere and the extraterrestrial solar irradiance. Appropriate inversion of these data products yields global information about the atmospheric absorption and scattering processes enabling the abundance of trace constituents such as O₃ and NO₂ to be retrieved.

The GOME measurements have been used in this study to determine the NO₂ tropospheric column distribution on a global scale. The retrieval of these unique data is just now possible with the measurements of the GOME instrument. In addition, two three-dimensional global chemistry transport models IMAGES (Intermediate Model for the Global Evolution of Species) [Müller and Brasseur, 1995; Pham et al., 1995; Granier et al., 1996, 1999, 2000a, 2000b] and MOZART (Model for Ozone and Related Chemical Tracers) [Brasseur et al., 1998; Hauglustaine et al., 1998] have been used to investigate the retrieved tropospheric NO₂ column amounts and help to interpret it. Both models use the EDGAR emission database [Olivier et al., 1996] for anthropogenic emissions and the database developed by Müller [1992] for natural emissions. The main differences between these models are the source of the atmospheric wind fields (monthly mean climatological fields for IMAGES and 6 hourly fields from a global circulation model for MOZART) and the representation of the boundary layer. MOZART has a less diffusive boundary layer than IMAGES, resulting in a more realistic representation of the physical and chemical processes in the lowest layers of the model.

In this decade a number of satellite instruments will be launched into low Earth orbit whose primary objective is to investigate the trace gas distributions in the lower troposphere, e.g., SCIAMACHY on ENVISAT [Bovesmann et al., 1999, and references therein], GOME-2 on METOP, Ozone Monitoring Instrument (OMI), and Tropospheric Emission Spectrometer (TES) on EOS-CHEM. An adequate understanding of atmospheric radiative transfer, in particular, in the boundary layer, is required to interpret accurately the slant column amounts from these nadir sounding instruments.

In this study the global distribution of tropospheric NO₂ column amounts retrieved from GOME level 1 data products have been compared with NO₂ column amounts inferred from three-dimensional chemistry transport models. Investigating the similarities and differences in the magnitude, global distribution, and seasonal variations of the retrieved and simulated NO₂ provides novel insight into the global behavior of NO_x. Importantly, the GOME tropospheric NO₂ columns reveal "hotspots" of significance for the assessment of regional, transnational, and intercontinental air pollution.

The algorithms used to retrieve NO₂ column amounts from GOME observations are discussed in section 2. Section 3 describes the chemistry transport models IMAGES and MOZART. In section 4 modeled (IMAGES and MOZART)

and observed (GOME) NO₂ column amounts are compared. The contributions of the different atmospheric model layers to the total tropospheric NO₂ column amounts and the different factors affecting the simulated NO₂ columns are discussed in section 5, followed by a comparison of the modeled NO₂ data with observations in section 6. The effect of the NO₂ columns on the radiative forcing of climate is probed in section 7, and section 8 summarizes the major conclusions to be drawn from our study.

2. Tropospheric NO₂ Columns Retrieved From GOME

In this section, the GOME instrument and its measurements are described, the retrieval methods applied to the GOME observations are discussed, and the resultant monthly averaged tropospheric NO₂ vertical column densities (VCDs) for 1997 are presented. As a large part of the tropospheric NO₂ column over polluted areas is believed to be located in the boundary layer, the sensitivity of GOME data to uncertainties in the retrieval of NO₂ in the boundary layer is considered and an estimate of the total tropospheric NO₂ VCD made, which accounts for clouds covering its source regions during the GOME overpasses.

Tropospheric NO₂ plumes typically occur on regional and continental scales, as shown in Plates 1c and 2 where locally large amounts are observed over Europe, eastern North America, parts of Asia, and the Johannesburg metropolitan area in South Africa. More than 90% of the tropospheric NO₂ is believed to be emitted into continental air, while the remaining 10% is generated over the ocean by lightning, and ship or aircraft emissions [e.g., Valks and Velders, 1999]. In contrast, stratospheric NO_x results largely from the reaction of nitrous oxide (N₂O) with excited oxygen atoms. In addition, the relatively short lifetime of NO_x in the troposphere, compared to that in the stratosphere, leads to its tropospheric distribution appearing as plumes emanating from source regions.

Over populated regions of the continents the majority of the NO₂ column is released from and believed to be located primarily in the boundary layer, with smaller amounts being transported to or generated in the free troposphere and stratosphere. In the remote maritime atmosphere the only significant source of tropospheric NO₂ is in-cloud lightning and transport from the continents. Analysis of both models of the stratospheric NO₂ behavior and the UARS HALOE NO₂ data products reveals that the NO₂ column in the upper atmosphere has significant longitudinal homogeneity up to and beyond latitudes of 65° for the same solar zenith angle. This is explained by the fact that the majority of this NO₂ is located in the middle and upper stratosphere, where it is relatively unperturbed by local changes in the troposphere, but strongly influenced by the solar zenith angle. In addition, advective mixing of NO_x at such altitudes is relatively rapid compared to the rate of its chemical production or loss.

The assumptions that the upper atmospheric column of NO₂ is longitudinally homogeneous and that tropospheric amounts of NO₂ above the clean remote ocean are small provide the physical basis enabling the tropospheric and stratospheric NO₂ columns to be separated from the inversion of nadir observations of GOME or SCIAMACHY. Two related but independent approaches have been used in this study. They both utilize the GOME level 1 data products: the

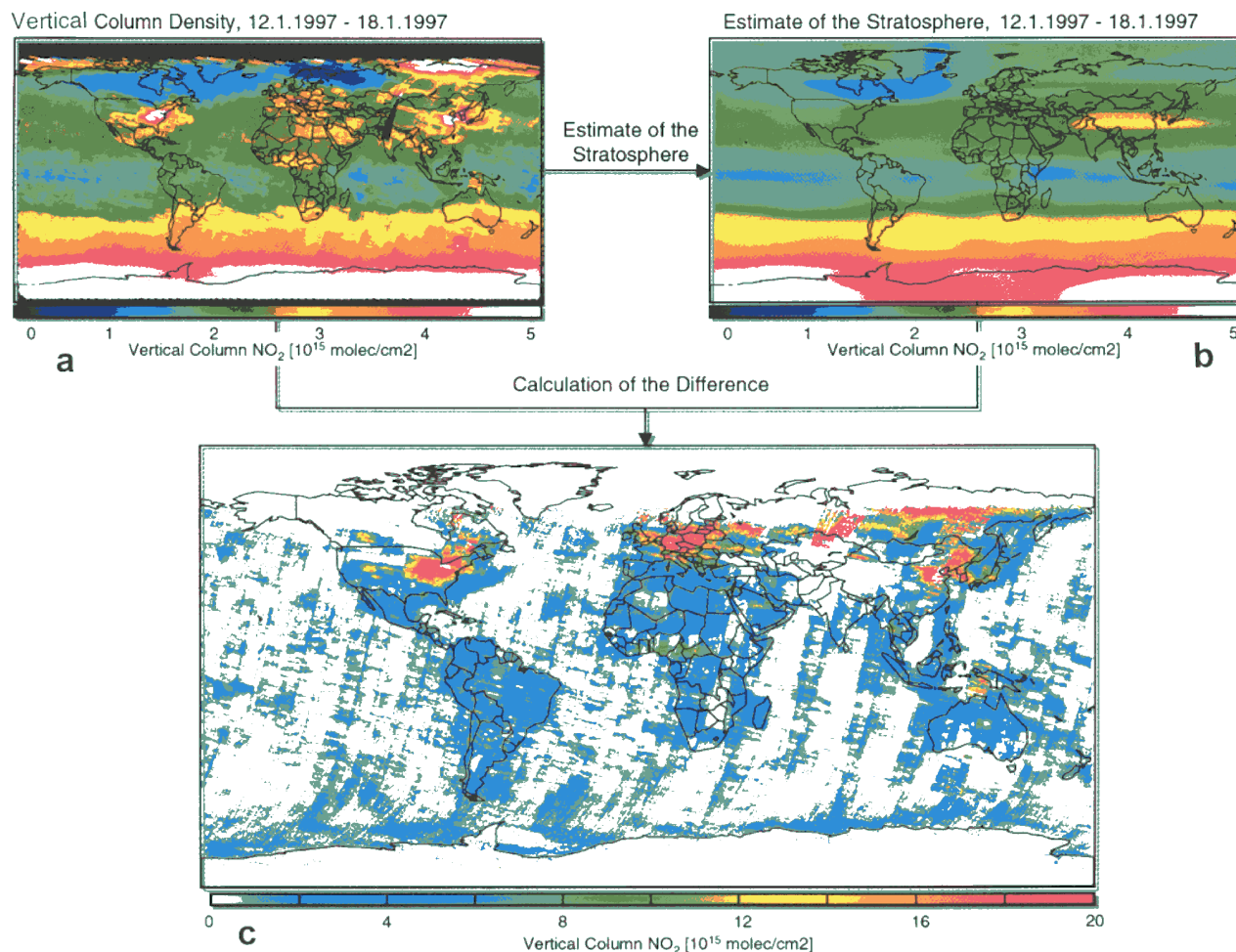


Plate 1. Flowchart of the steps in the algorithm to separate the stratospheric and tropospheric parts of the GOME NO₂ column. (a) the land masses and cloudy pixels are masked out and (b) the stratosphere is estimated using a normalized convolution. (c) The tropospheric residual is then estimated by calculating the difference between Plates 1a and 1b. Plate 1c shows the GOME tropospheric NO₂ column with the correct air mass factors depending on solar zenith angle, ground albedo and clouds. Note the different scales of the maps.

top of the atmosphere radiance (TOARAD), observed from the GOME viewing geometry, and the extra-terrestrial solar irradiance (ETSIR).

2.1. GOME Instrument and Observations

The GOME instrument, on board the second European Research Satellite (ERS-2) comprises a scanning mirror, spectrometer, thermal and electronic subsystems [Burrows *et al.*, 1991, 1993, 1999]. Light from the atmosphere is collected by the scan mirror then focused on the entrance slit of the spectrometer by an off-axis parabolic telescope. After being collimated, the light passes through a predispersing prism forming an intermediate spectrum within the instrument. The predispersing prism is such that internal reflection results in a polarized beam, which is directed towards the polarization monitoring device (PMD). The latter comprise three broadband detectors, which observe upwelling radiation from the atmosphere polarized relative to plane within the instrument in the wavelength ranges 300–400 nm, 400–600 nm, and 600–800 nm, respectively. The main beam from the predispersing prism forms a spectrum within the instrument.

Different sections of this spectrum are then directed toward the four spectral channels of GOME: each channel comprising a grating, transmissive optics and a 1024 element diode array. In this manner the entire spectrum between 232 and 793 nm may be observed: the spectral resolution being ~ 0.2 and 0.33 nm, below and above 400 nm respectively.

As GOME is optimized for the collection of the upwelling radiation from the atmosphere directly by the scan mirror, it is necessary to direct the extra terrestrial solar output over a diffuser plate to reduce its intensity prior to it being reflected by the scan mirror into the instrument. In this manner both the upwelling radiance and the extraterrestrial irradiance are measured by GOME. The signals, recorded by GOME, are dumped to the ESA ground station at Kiruna. These data are processed for ESA by the GOME Data Processor at the German Atmospheric and Space Research Processing and Archiving Center (DLR-PAC): the resultant level 1 products, TOARAD and ETSIR, and level 2 total ozone column data product are subsequently being distributed to the scientific community.

ERS-2 was launched on the April 20, 1995, into a sun

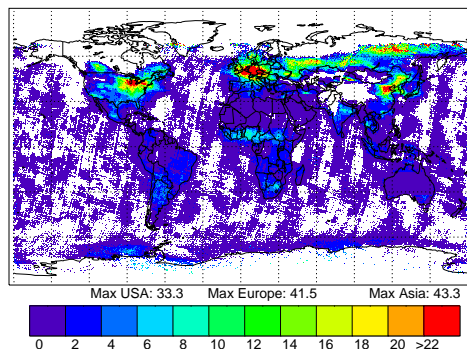
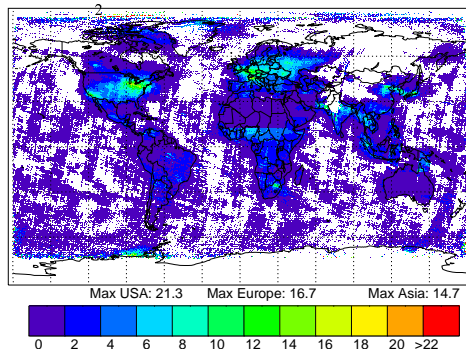
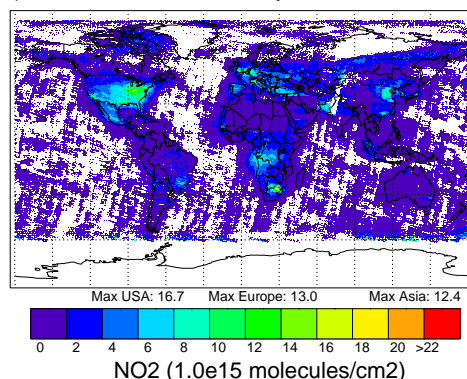
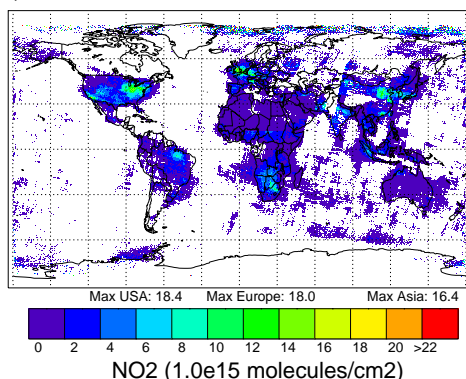
a) GOME: NO₂ column: January 1997b) GOME: NO₂ column: April 1997c) GOME: NO₂ column: July 1997d) GOME: NO₂ column: October 1997

Plate 2. Monthly averaged tropospheric NO₂ columns as derived from GOME measurements for (a) January, (b) April, (c) July, and (d) October 1997. The measurements are made around 1030 LT. The grid shown is 0.45° x 0.45° and derived from both cloudy and cloud free pixels. Blank pixels indicate that there is no data. The maximum values that occur over the United States, Europe, and Asia in a single grid cell are given below the plots. The maximum GOME NO₂ values on the IMAGES grid are (21.9, 30.6, 17.8) for (United States, Europe, Asia) for January, (13.1, 11.0, 6.5) for April, (12.2, 8.7, 7.7) for July, and (12.6, 9.8, 7.9) for October (all in 10¹⁵ molecules cm⁻²).

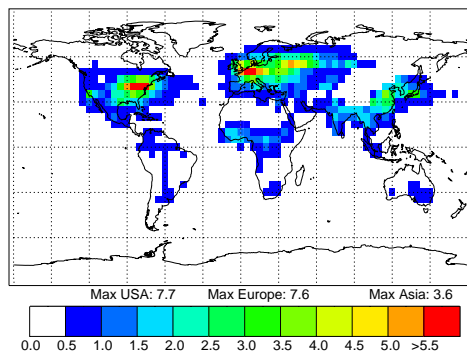
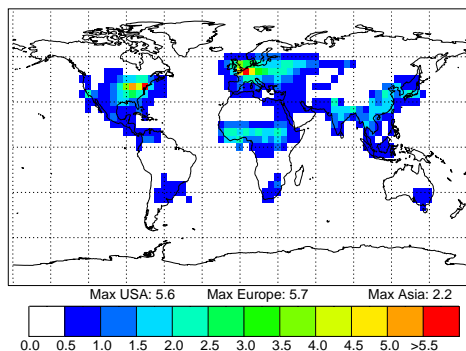
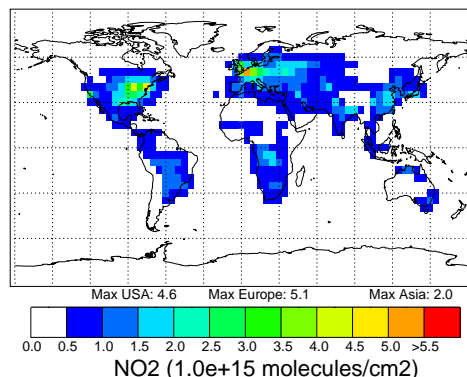
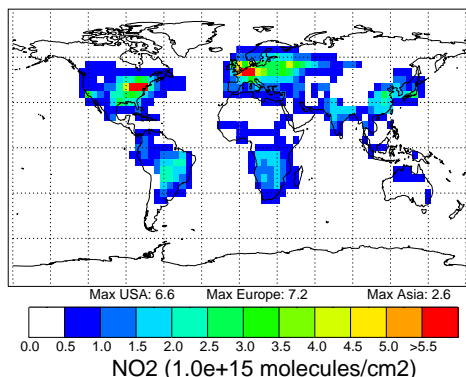
a) IMAGES: NO₂ column: Januaryb) IMAGES: NO₂ column: Aprilc) IMAGES: NO₂ column: Julyd) IMAGES: NO₂ column: October

Plate 3. Monthly averaged tropospheric NO₂ columns as calculated by IMAGES for (a) January, (b) April, (c) July, and (d) October. Data correspond to ~1030 LT. The maximum values that occur over the United States, Europe, and Asia in a single grid cell are given below the plots. Note the color scale is 4 times smaller than the scale used for the GOME plots and the lowest level (0-0.5) has been left blank to obtain clearer plots.

synchronous orbit; having an equator crossing time of 10:30 am in a descending node. The scan strategy of GOME yields complete global coverage at the equator in 3 days: the swath width of GOME being 320 x 40 km² for the array detector and 20 x 40 km² for the PMDs, which are read out 16 times faster than the arrays. More details about the design and the first results from GOME are provided elsewhere [Burrows *et al.*, 1999, and references therein].

2.2. Retrieval Algorithm for Tropospheric NO₂

Using the differential optical absorption spectroscopy (DOAS) technique [e.g., Platt, 1994], it has previously been shown that the total column amounts of ozone (O₃) [Burrows *et al.*, 1998c] bromine monoxide (BrO) [Hegels *et al.*, 1998; Richter *et al.*, 1998; Wagner and Platt, 1998], chlorine dioxide (OCIO) [Burrows *et al.*, 1998c; Eisinger *et al.*, 1996; Wagner *et al.*, 1999; Burrows *et al.*, 1999], formaldehyde (HCHO) [Burrows *et al.*, 1999] sulfur dioxide (SO₂) [Eisinger and Burrows, 1998, and references therein], nitrogen dioxide (NO₂) [Burrows *et al.*, 1998c; Leue *et al.*, 1999], and water vapor (H₂O) [Noel *et al.*, 1999] may be retrieved from GOME irradiance and radiance measurements. In addition ozone and aerosol profiles may also be retrieved [Hoogen *et al.*, 1999, and references therein].

In this study the DOAS spectral window chosen for NO₂ retrieval from GOME is 425-450 nm [Wagner, 1999]. Preflight measurements of the absorption cross section of NO₂, O₃ [Burrows *et al.*, 1998a, 1999], H₂O, and O₄ and calculations of an effective Ring spectrum [e.g., Bussemer, 1993; Vountas *et al.*, 1998] serve as input for the retrieval. The NO₂ slant column densities derived by DOAS are converted into total atmospheric NO₂ vertical column densities (VCDs) by applying a calculated air mass factor (AMF) for each observation.

The AMF describes the weighted path of the light through the atmosphere compared to the vertical path. In general, the AMF is a function of the solar zenith angle, the selected wavelength interval, the vertical profile of NO₂, amounts of aerosol, the surface spectral reflectance or albedo, the cloud cover, and absorption by the atmosphere. AMFs for this study have been calculated with the radiative transport models AMFTRAN [Marquard *et al.*, 2000] for the total atmosphere and GOMETRAN [see Rozanov *et al.*, 1997] for the troposphere. The albedo used for AMF calculations in this study are those derived using by cloud free scenes, and an atmospheric model [Richter and Burrows, 2000].

In order to retrieve the tropospheric column amounts of NO₂ from GOME data, a priori knowledge about the behavior of the stratospheric NO₂ is required. It is assumed that stratospheric NO₂ is longitudinally homogeneous and the tropospheric NO₂ in the clean remote troposphere above the ocean is negligible. In one approach, known as the tropospheric excess method (TEM) [Burrows *et al.*, 1999], tropospheric NO₂ in the clean remote troposphere over the oceans around 180° longitude is assumed to be negligible. Average three days, weekly, monthly, and yearly global maps can then be derived. In the TEM approach the percentage of cloud cover within a scene is estimated using the higher spatially resolved broad band measurements by a threshold algorithm [Burrows *et al.*, 1998b]. In this manner, composite maps, typically between 60°N and 60°S, showing the tropospheric column of NO₂ for different percentages of clouds can be obtained.

Image-processing techniques (IPT) applied to 3-day composite images of the GOME NO₂ observation has been used extensively in this study. Plate 1 illustrates schematically the individual IPT steps used to discriminate between the stratospheric and tropospheric contribution to the total NO₂ VCDs. For the marine regions the following procedure was used. First, the continents plus a band ~200 km off shore is masked out to avoid regions of NO₂ emissions on and near coastlines. The stratospheric column of NO₂ VCDs is assumed to dominate the resultant region. A normalized convolution is then applied to the data (see Jähne *et al.* [1999] for details), enabling the global stratospheric NO₂ distribution to be inferred. For the IPT algorithm, marine NO₂ measurements for clear sky conditions were selected for which radiative transport calculations can be performed very precisely. For this selection the cloud fraction is derived from the onboard-operated PMD instrument as described by Wenig *et al.* [1999].

Plate 1b shows the “IPT stratospheric background” thus derived. The resulting tropospheric NO₂ VCDs were then obtained from the differences in total atmospheric and inferred stratospheric NO₂ VCDs. A more detailed description of the IPT retrieval of tropospheric NO₂ VCD from the total atmospheric NO₂ can be found in the works of Leue [1999], Leue *et al.* [1999, 2001], and Richter and Burrows [2000].

In the IPT algorithm, the difference between the total (SCD_{total}) and the stratospheric NO₂ slant column densities (SCD_{strat}) is attributed to tropospheric NO₂. The tropospheric NO₂ VCDs are then computed:

$$\text{VCD}_{\text{trop}} = (\text{SCD}_{\text{total}} - \text{SCD}_{\text{strat}}) / \text{AMF}_{\text{trop}}$$

The tropospheric air mass factors (AMF_{trop}) were derived using the radiative transfer model GOMETRAN [Rozanov *et al.*, 1997], assuming a well-mixed boundary layer of average height 1.5 km, containing the entire tropospheric NO₂ layer, an average surface albedo being 5%, and a maritime aerosol distribution assumed to be located in the boundary layer.

It should be noted that the results of the IPT algorithm are in good agreement with those of an other approach, known as the tropospheric excess method (TEM) [Burrows *et al.*, 1998c, 1999; Richter and Burrows, 2000]. The main difference between TEM and IPT is that in the TEM algorithm cloud free pixels are selected, while in the IPT algorithm all pixels are used and cloud effects are taken into account a posteriori. Because a cloud threshold value is used in the TEM analysis, the residual cloud contamination will lead to an underestimation of tropospheric NO₂.

2.3. GOME's Sensitivity to Boundary Layer NO₂ Observations

The relative penetration of photons in the boundary layer, as observed at the top of the atmosphere, depends strongly on factors including the cloud cover, surface spectral reflectance (albedo), and tropospheric aerosol loading. Thus the weighting of the measurement sensitivity to the lowest levels in the troposphere is a strong function of the wavelength and viewing conditions. This results in a variable sensitivity of the retrieved DOAS NO₂ absorption to the NO₂ in the lowest layer of the atmosphere, the boundary layer. For example, calculations using GOMETRAN show that for cloud free conditions and overhead sun roughly 20% of the incoming solar photons at 437 nm are backscattered by atmospheric molecules before reaching the surface. The backscattering by

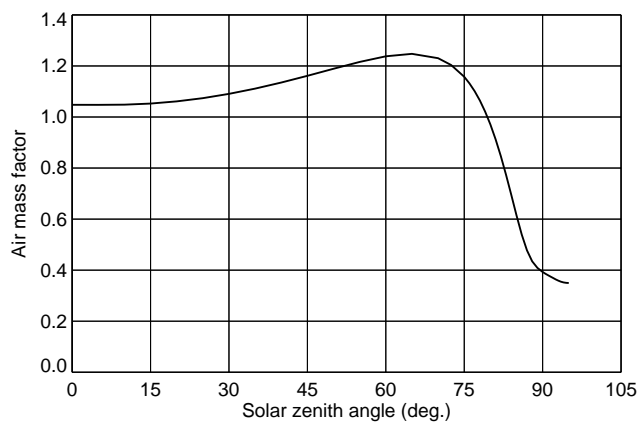


Figure 1. Air mass factor of NO₂ for a boundary layer of 1.5 km at a wavelength of 437.5 nm, assuming an albedo of 5% and a maritime aerosol layer in the boundary layer.

aerosol varies in the range of 1-10% and is dependent on type and distribution. As a result of this sensitivity to the surface spectral reflectance (albedo) in the blue spectral range, ~2-12% of the remaining 80% of the total incoming photons are reflected from nonsnow covered land surfaces, and 4-8% from the oceans [Richter and Burrows, 2000]. This behavior

is reflected in the air mass factor (Figure 1), which is small and nearly independent of solar zenith angles below 80°.

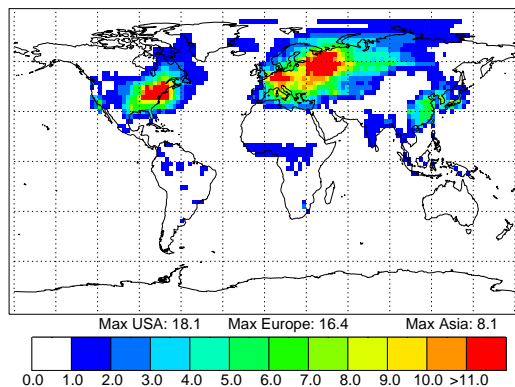
Our AMF calculations also show that strongly absorbing aerosols located in the boundary layer further obscure the visibility of boundary layer NO₂ for space-borne observations. Multiple Mie scattering by aerosol or cloud particles, however, can also increase the tropospheric absorption due to increasing optical path length. The AMF depends strongly on the surface spectral reflectance, for example increasing the surface spectral reflectance from 4 to 8% increases the tropospheric AMF by ~30%.

Optically thick clouds effectively obscure tropospheric NO₂ located below the cloud top because they act as highly reflecting surfaces, having typically an albedo in the range 80-95% [Kurosu *et al.*, 1997]. As a result of their brightness, even a small cloud fraction within a GOME pixel may contribute significantly to the signal received by the instrument, decreasing the sensitivity toward the detection of boundary layer NO₂, but increasing the relative weighting of NO₂ above the cloud to that of the free troposphere or stratosphere.

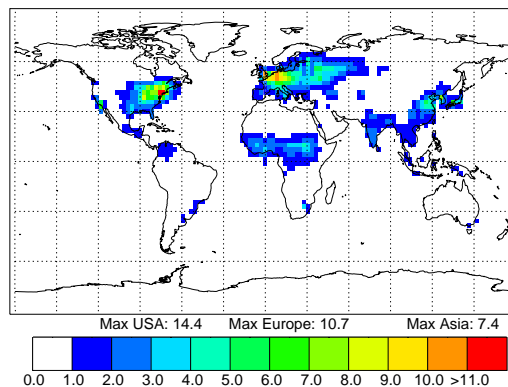
2.4. Estimation of the Total NO₂ Column

In this study, in order to estimate the total tropospheric NO₂ VCD all GOME observations (including also the cloudy regions) were taken into account. This procedure enhances the

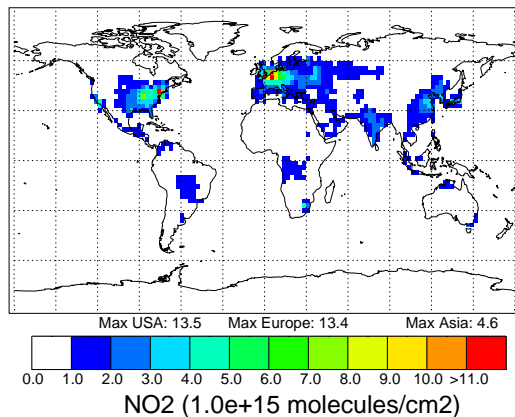
a) MOZART: NO₂ column: January



b) MOZART: NO₂ column: April



c) MOZART: NO₂ column: July



d) MOZART: NO₂ column: October

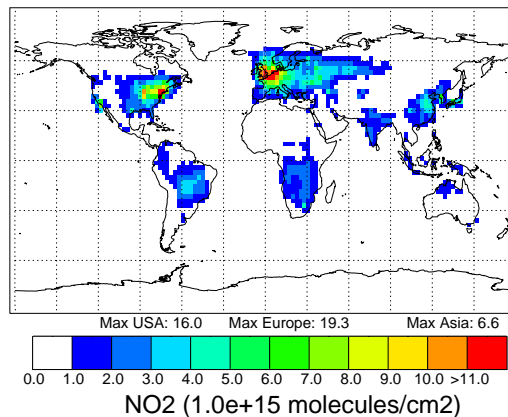


Plate 4. Monthly averaged tropospheric NO₂ columns as calculated by MOZART for (a) January, (b) April, (c) July, and (d) October. Data shown correspond to ~1030 LT. The maximum values that occur over the United States, Europe, and Asia in a single grid cell are given below the plots. Note the color scale is 2 times smaller than the scale used for the GOME plots and the lowest level (0-1) has been left blank to obtain clearer plots.

total number of considered measurements but requires a correction for the effects of clouds on the retrieved tropospheric NO₂. While, in principle, this can be done for individual GOME measurements, we here apply an average correction for monthly averaged GOME measurements of tropospheric NO₂. This approach was chosen in order to reduce the complexity of the correction procedure. In addition, there are still considerable uncertainties in the determination of the individual cloud fractions using current methods.

The average correction applied here makes use of assumptions of cloud fraction and albedo which are described below. They are consistent with conditions over NO₂ source regions, including in particular industrialized regions, but are considered to inappropriate to other regions, e.g., deserts or snow covered regions. Thus the comparisons in this study focus on GOME tropospheric columns of NO₂ and the model results for regions where the above assumptions are valid.

From GOME measurements it turned out that the total atmospheric reflectance (including scattering on the surface, and on aerosols and molecules) is ~25% at 437 nm, whereas the albedo above clouds is equal to or greater than 80%. Thus the geometric cloud fraction has to be weighted by these relative intensities to determine the sensitivity of GOME for the troposphere (assuming that the tropospheric NO₂ column is entirely below the clouds). For a geometric cloud fraction of 0.5 the relative contribution of the clear sky part of the GOME pixel to the total measured signal is ~24%. Thus the respective tropospheric NO₂ column density has to be corrected by about a factor of 4.2. For a geometric cloud fraction of 0.4 (0.6) the relative contribution of the clear sky part to the total signal is ~31% (17%), resulting in correction factor of ~3.2 (5.8). Assuming a Gaussian frequency distribution of the geometric cloud fractions of GOME ground pixels (with a maximum at a cloud fraction of 0.5 and a standard deviation of 0.25), we then derive an average correction factor of about 4 which was applied to tropospheric NO₂ derived by the IPT algorithm, including all the cloudy pixels. No correction factor was applied to the Saharan, Arabian Peninsula, and Australian deserts, since these areas are almost always cloud free.

It should be noted that the assumed frequency distribution of the cloud fractions for GOME ground pixels is somewhat arbitrary but should provide suitable values for many parts of the world (including, e.g., industrialized regions). Future work will yield more precise information on statistics of GOME cloud fractions from improved GOME cloud determination algorithms.

Comparing the TEM NO₂ VCD for all scenes with those having 30% cloud cover, indicates that the TEM NO₂ VCD is typically a factor of 2 larger for the 30% case than over industrial source regions, where the majority of NO₂ is emitted. These findings are in fair agreement with the average cloud correction applied to the NO₂ data derived from the IPT algorithm as described above.

The uncertainty in the tropospheric NO₂ columns from GOME depends mainly on four factors. (1) The precision of the NO₂ DOAS fit, which is limited by photon and electronic noise and can become relatively large for a single observation. Nevertheless, for monthly averaged values this error decreases to very small values ($\pm 1\%$) and can thus be neglected. (2) Systematic errors in the NO₂ DOAS fit (caused, for example, by uncertainties in the wavelength calibration of the reference

spectra) do not decrease if several observations are averaged. However, by subtracting the stratospheric column from the total column most of the possible systematic errors cancel out; the remaining systematic error is estimated to be $\pm 10\%$. (3) The most important uncertainties come from the modeling of the tropospheric AMFs. The lack of precise information about the ground albedo and the tropospheric NO₂ height profile cause errors in the calculated tropospheric NO₂ VCDs which can be up to ~50% for a single observation. For monthly averages these errors partly cancel out leaving an uncertainty of ~15%. (4) The uncertainties caused by clouds constitute the largest contribution to the error. For a single observation it might be up to ~100%. For the monthly averages these errors partly cancel out; we estimate the remaining uncertainty to be ~50%. This last uncertainty dominates the total error in the retrieved tropospheric NO₂ VCDs.

2.5. Monthly Averaged NO₂ Column Amounts for 1997

Plate 2 shows the inferred tropospheric NO₂ columns (=VCDs) for the months of January, April, July, and October in 1997. As expected, the highest NO₂ columns occur over industrialized areas in the Northern America, Europe, and China but hotspots having localized pollution in South Africa, the Arabian Peninsula, and Japan are also clearly discernable. Enhanced tropospheric NO₂ columns also occur in known regions of biomass burning and savannah fires in South America, Africa, and Indonesia (July and October). Much lower NO₂ columns are observed for remote continental regions in Australia or the Sahara desert. The inferred tropospheric NO₂ columns for the remote oceans are close to zero, mainly because for this area the total atmospheric NO₂ column was assumed to be mostly located in the stratosphere (see above). In January the maximum NO₂ columns for the eastern United States, western Europe, and eastern China are twice as large as in the other months; i.e., $33\text{--}43 \times 10^{15}$ molecules cm⁻² versus $14\text{--}20 \times 10^{15}$ molecules cm⁻². The larger winter values likely result from a decreased loss of NO₂ by reaction with OH (the major NO_x loss process in the lower troposphere), confinement in the boundary layer, reduced vertical transport from the boundary layer to the free troposphere, and slightly increased source strengths.

3. Description of the Models

3.1. IMAGES

The IMAGES model [Müller and Brasseur, 1995; Pham *et al.*, 1995; Granier *et al.*, 1996, 1999, 2000a, 2000b] is a three-dimensional global chemistry transport model that calculates monthly averaged distributions of some 60 species. It has a horizontal resolution of 5° x 5° and includes 25 levels in the vertical extending from the surface to 50 hPa (~22 km). The dynamics is governed by prescribed monthly averaged transport and temperature fields. The distribution of clouds and precipitation rates are also prescribed according to monthly averaged climatological values. The rapid turbulent exchanges in the planetary boundary layer are parameterized using a diffusion coefficient depending on the vertical gradient of potential temperature. The coefficients representing this diffusion coefficient are calculated using monthly mean temperature data.

The chemical scheme in the model is adapted from Müller and Brasseur [1995], and its current version is given by

Granier et al. [2000b]. The chemistry includes an explicit description of nonmethane hydrocarbons chemistry including the formation of peroxy-acetyl nitrate (PAN), important for the transport of reactive nitrogen from source regions to the free troposphere [*Moxim et al.*, 1996; *Horowitz and Jacob*, 1999]. The reaction rates are from *DeMore et al.* [1997]. The anthropogenic emissions are from the EDGAR emission database [*Olivier et al.*, 1996], which is developed in close cooperation with the Global Emissions Inventory Activity (GEIA) of the International Global Atmospheric Chemistry/International Geosphere-Biosphere Programme IGAC/IGBP [see also *Granier et al.*, 2000b]. These emissions are specified on a 1° x 1° global resolution for CH₄, NO_x, CO, and nonmethane hydrocarbons. They include emissions from fossil fuel use (industry, transport, fuel production, and transmission), biofuel combustion, industrial processes and solvent use, waste treatment, and agricultural waste burning. A seasonal variation is imposed on the anthropogenic emissions according to *Müller* [1992]. Biomass burning emissions are taken from *Granier et al.* [2000a], based on inventories by *Hao and Liu* [1994]. The production of nitric oxide by lightning discharges is distributed according to the flash frequencies obtained from satellite measurements as reported by *Turman and Edgar* [1982]. Total emission of NO from lightning is assumed to be 5 Tg N/yr.

IMAGES calculates a full diurnal cycle in the first 3 days of the month, with time steps of 1 hour for the first 2 days and a time step of 0.5 hour for the third day. For the rest of the month, diurnal average values are calculated and a time step of 6 hours is used.

3.2. MOZART

The MOZART-2 model is a new version of the model described by *Brasseur et al.* [1998] and *Hauglustaine et al.* [1998]. It is a three-dimensional global chemistry transport model which can calculate the distribution of ~60 species, with a 20 min time step. It has a horizontal resolution of 2.8° x 2.8° and 34 levels in the vertical extending from the surface to 4 hPa (~36 km). The meteorological parameters needed to calculate the advective transport, smaller-scale exchanges, and wet scavenging of long lived chemical species are prescribed and taken from the output of the middle atmosphere version of the National Center of Atmospheric Research (NCAR) community climate model CCM3 and are updated every 6 hours. The parameterization of exchanges in the boundary layer is based on *Holtstlag and Boville* [1993] and uses a vertical eddy flux proportional to an eddy diffusion coefficient. The latter depends on a turbulent velocity scale and on the boundary layer height (Richardson number dependent). A countergradient term representing nonlocal transport associated with dry boundary layer convection is also taken into account.

The chemical scheme in the model is the same as the one used in IMAGES [*Granier et al.*, 2000b], with reaction rates from *DeMore et al.* [1997]. The emissions of CH₄, CO, NO_x, and hydrocarbons are the same as those used in IMAGES (see section 3.1). The production of NO by lightning is distributed as a function of space and season according to the location of convective clouds as provided by the general circulation model, and the height of the top of convective clouds, as formulated by *Price and Rind* [1992]. As for the IMAGES model, the global production of NO from lightning is assumed to be 5 Tg N/yr. For our calculations only diurnal

average concentrations of NO₂ were stored, calculated from the mixing ratios at every 20 min time step. The intermediate data were not saved separately.

4. Modeled NO₂ Column

In the Plates 3 and 4 the modeled tropospheric NO₂ columns (vertical column densities) calculated by the IMAGES and MOZART models (respectively) are shown. Plates 3 and 4 can be compared with Plate 2 showing the GOME NO₂ columns. To obtain a clearer picture, the lowest levels (0-0.5 or 0-1 x 10¹⁵ molecules cm⁻²) are left blank. The modeled tropospheric NO₂ columns are obtained from the calculated NO₂ mixing ratios by integrating from the surface to the tropopause.

GOME is in a Sun synchronous orbit. It passes over the equator at 1030 local time (LT) and covers all longitudes at the equator in 3 days. To agree as closely as possible with the local overpass time of GOME, the NO₂ columns as calculated by IMAGES correspond with a local time between 1000 and 1040 LT. They are derived by calculating the ratio of the NO₂ concentration at ~1030 LT and the diurnal average concentrations in the full diurnal cycle of the model. The diurnal average concentration in the middle of the month is then multiplied by this ratio. Details about the difference between the column near 1030 LT and the diurnal average are discussed in section 5 and Plate 7. For the MOZART calculations only diurnal average concentrations of NO₂ were stored, no hourly data. To calculate the NO₂ column amounts from MOZART at the GOME overpass time, we used the results from the IMAGES model to calculate the ratio between the tropospheric NO₂ columns at 1030 LT and the diurnal average columns. This ratio was then used, on a monthly basis, to calculate the columns at 1030 LT for the MOZART calculations.

The NO_x emissions in the model calculations are monthly or yearly averages. Diurnal variations are not taken into account. About 30% of the global NO_x emissions come from road transport. Especially in and around the bigger cities road transport exhibits a strong diurnal variation. During the morning rush hours, larger emissions of NO_x are expected which may affect the NO₂ column at the GOME overpass time (1030 LT). We have performed a calculation with IMAGES using a diurnal variation in the NO_x emissions from road transport according to *Dreher and Harley* [1998]. The effect of a diurnal variation in NO₂ emissions (versus a constant emission) on the tropospheric NO₂ column was found to be small, with very local increases around Los Angeles of ~3%, in the eastern United States 1-5%, northwestern Europe 1-4%, and Japan ~1% (annual averages).

4.1. Grid Sizes and Method of Calculation

Before comparing the modeled NO₂ columns with the GOME measurements, the effect of the different grid sizes has to be considered. In Plates 5a and 5b the GOME NO₂ columns are shown mapped onto the IMAGES and MOZART grids. This smoothes the plots and removes the small very localized spots with high NO₂ columns over industrialized areas. In Plate 5 the corresponding annual averaged modeled NO₂ columns are also shown. In the middle panels the tropospheric columns are calculated by integrating from the surface

to the tropopause, while in the lower panels a similar method as for the GOME columns is applied; i.e., subtracting the NO₂ column over the Pacific (around 180°W) from the NO₂ column at all longitudes. The latter procedure is performed on a monthly basis and retains the latitudinal variation. Comparing both procedures (Plate 5c versus 5e and 5d versus 5f), it can be seen that there is virtually no difference for industrialized areas with large tropospheric NO₂ columns as well as for the more remote continental areas (South America, Africa, and Australia). There are some differences over the oceans, but they are small in an absolute sense. The tropospheric NO₂ columns over the oceans and away from NO_x sources are smaller than 0.5×10^{15} molecules cm⁻² (middle panels). This is the amount that is subtracted from both the GOME (Plates 5a and 5b) and modeled (Plates 5e and 5f) in the derivation of the tropospheric NO₂ columns but is part of the tropospheric column.

4.2. Comparing Models With GOME

Plates 3 (IMAGES) and 4 (MOZART) can now be compared with Plate 2 (GOME). We must be cautious in making a quantitative comparison, considering the large uncertainties in the retrieval of the tropospheric NO₂ columns from the GOME measurements as discussed above (sections 2.3 and 2.4); these are likely to be as large as 50% for individual locations. A qualitative comparison of GOME with the model calculations is, however, useful, since this uncertainty is likely to affect the NO₂ columns everywhere in a similar manner. Note that Plates 2, 3, and 4 have different scales to show the full range of the NO₂ columns. The figures also differ in the grid sizes (grid size of the GOME data is 0.45° x 0.45°). The magnitude of the NO₂ columns varies greatly in different regions. Because of the color scale chosen, the low NO₂ columns over remote areas are not easily visible in these plates. In Plate 5 the columns from GOME are plotted on the IMAGES and MOZART grids, and the color scale is nonlinear to accentuate the areas with small columns.

The overall picture of the measured tropospheric NO₂ columns is well reproduced by the model calculations: large columns over industrialized areas both in the Northern and Southern Hemispheres, low values over remote continental areas, and very low values over the oceans. Larger tropospheric NO₂ columns are found in winter than in summer, the same as in the GOME observations. The absolute values of the tropospheric NO₂ columns as calculated by MOZART are close to those derived from the GOME measurements, while the column amounts from IMAGES are significantly smaller. The maximum values in IMAGES are about a factor of 2-3 smaller than in GOME for the United States, ~1.5-4 smaller for Europe, and ~3-5 for Asia, using the same grid (that of IMAGES) for both. The maximum values in MOZART are similar to those of GOME for the United States (a factor of 0.8-1.5 smaller) and Europe (0.7-2), but a factor of 1.5-3.5 smaller for Asia (all on the MOZART grid). If averaged over a larger area (subcontinental), the high maximum values in very localized areas in MOZART and in GOME are somewhat reduced. The agreement between the model calculations and the GOME measurements is somewhat worse for South America and Africa, areas with lower NO₂ columns in which biomass burning, savannah fires, and soil emissions are the dominant NO_x sources. Uncertainties in the structure of the boundary layer due to larger convective activity in the tropics

may also affect the model values in these regions. The tropospheric NO₂ columns in IMAGES are about a factor of 2-5 smaller than GOME for South America and Africa. The values in MOZART are ~1-2 times smaller for South America and ~3 times for Africa. The modeled columns over the Saharan and Australian deserts are small ($0.2-0.5 \times 10^{15}$ molecules cm⁻²).

The following points have to be mentioned in relation to the comparison: (1) The emissions used here correspond with a 1990 inventory and are probably too low for 1997, especially for China. *Kato and Akimoto* [1992] found an increase in NO_x emissions in eastern Asia of 4%/yr in the 1970s and 1980s. Extrapolating this trend gives an increase in NO_x emissions for this area of 32% between 1990 and 1997. This can explain only part of the difference between GOME and MOZART. A larger increase in emissions would be needed to explain the whole difference. (2) Differences for South America and Africa can also be caused by errors in the parameterization of convective processes which transport NO₂ from the boundary layer to higher altitudes. (3) In this comparison we only used one year of GOME data. The NO₂ columns probably display some year-to-year variability, but these real differences are likely to be smaller than the large uncertainties in the GOME tropospheric column amounts and uncertainties in the modeled representation of the boundary layer, suggesting that a better understanding of those factors will be needed before inter annual fluctuations can be meaningfully measured and compared to model calculations.

The differences between the measured and modeled columns can also be seen in Plate 6 where the ratios between the NO₂ column amounts are plotted, i.e., GOME/IMAGES and GOME/MOZART. For calculating these ratios the GOME data have been converted to the IMAGES or the MOZART grid. No ratio is calculated if either the measured or modeled NO₂ column was smaller than 0.1×10^{15} molecules cm⁻². Plate 6a shows that for most areas the GOME NO₂ column is 2-3 times larger than the column calculated by IMAGES. Comparing GOME with MOZART (Plate 6b) lower ratios (i.e., better correspondence) are found for the polluted areas; a factor of 0.5-2 for eastern United States and western Europe. The columns measured by GOME over the Sahara and Australia are 1.5-2 times larger than modeled by MOZART. The high ratios around 60°N over western Russia are the result of large NO₂ columns in GOME, probably resulting from an inadequate subtraction of the stratospheric NO₂ column from the total column for these areas.

From these comparisons between the GOME and modeled tropospheric NO₂ columns we conclude that there is a good qualitative agreement between the column amounts retrieved from the GOME measurements and both sets of model calculations. Quantitatively, the GOME columns are of the same order as the ones calculated by MOZART for the polluted industrialized areas.

The results of the IMAGES model differ substantially from those of the MOZART model. This difference is caused mainly by the different representations of the boundary layer and resulting different NO₂ profile in the lowest kilometer of the modeled troposphere in the two models. The MOZART model has a physically more realistic boundary layer with high and more or less homogeneous NO₂ concentrations. In IMAGES the boundary layer cannot maintain a high NO₂ concentration throughout the whole layer. Emitted pollutants are removed too fast from the boundary layer by vertical

diffusion. The boundary layer, which contributes strongly to the total tropospheric NO₂ column in polluted areas, therefore contains less NO₂ in IMAGES than in MOZART (see also section 6).

5. Contributions to the NO₂ Column

Now that we have considered the modeled tropospheric NO₂ columns in section 4, we probe the build up of the NO₂ column, some aspects of the chemical partitioning, and the contributions from different vertical layers based on model calculations only.

5.1. NO_x and NO_y

As mentioned above, GOME passes over the equator at 1030 LT, so it performs its measurements at various latitudes between ~10 and ~11 LT. In Plate 7a the difference in tropospheric NO₂ column at ~1030 LT and the diurnal average column is given as a ratio as calculated by IMAGES. Over industrial areas in the eastern United States and western Europe the tropospheric NO₂ column at the GOME overpass time is ~80% of the diurnal average column. Over South America and Africa it is slightly less, 50-70%. Any errors in the diurnal partitioning of NO₂ are therefore likely to be too small to substantially affect the comparison between GOME and the model calculations.

In the atmosphere, NO and NO₂ exhibit a clear diurnal variation: during the day NO₂ is converted to NO by photolysis, while during the night NO is converted back into NO₂ mainly by the reaction with ozone. The sum of NO and NO₂ (=NO_x) remains more or less constant during this cycle. The annual average lifetime of NO_x in the IMAGES troposphere is ~12 hours, while it is ~7 hours in the lowest 5 km of the Northern Hemisphere between 20°N and 70°N. It is interesting to see how large the tropospheric NO₂ column is compared with the tropospheric NO_x column. An error in the modeled NO₂/NO_x ratio could alter the comparison with GOME measurements. This ratio, yearly averaged, is shown in Plate 7b and 7c. In IMAGES (Plate 7b) the NO₂ column is 60-70% of the NO_x column over the continents (diurnal and annual average) with maximum values of ~80% for the eastern United States, western Europe, and eastern China. The NO₂/NO_x ratio for MOZART (Plate 7c) is overall slightly higher. Over the most polluted areas ratios of ~60% are found. These lower ratios are most pronounced in the winter months with values as low as 30%, while in summer, there is no difference over the United States and Europe (ratio ~80%). The lower NO₂ and higher NO values in winter are probably the result of direct NO emissions, slower photolysis of NO₂ to NO, and reduced conversion of NO to NO₂ because of lower ozone concentrations. This effect is not observable in the IMAGES model, probably because of the diffusive nature of the boundary layer, which transports NO and NO₂ upward out of the boundary layer too rapidly. An error in the NO₂/NO_x ratio could affect the agreement with the GOME measurements, although the modeled ratio is close to the observed ratio of ~80-90% in the boundary layer in eastern North America in August [Parrish *et al.*, 1993].

Assuming the total NO_x emissions are relatively well known, the tropospheric NO₂ column in the model calculations can also be affected by a different distribution of nitrogen species between NO₂ (or NO_x) and NO_y. The NO_y

species with the largest tropospheric column is HNO₃, with ~10-18 x 10¹⁵ molecules cm⁻² over the industrialized areas for both IMAGES and MOZART. The NO_x/NO_y ratio over the industrialized areas in the United States and Europe is ~20-40% in IMAGES and 30-70% in MOZART. The observed NO_x/NO_y ratio [Parrish *et al.*, 1993] ranges from ~30 to ~50% at daytime and from ~50 to ~70% diurnally averaged in eastern North America in August. The MOZART values are in better general agreement with these observations than the IMAGES values. The models differ on their treatment of the nitrogen species resulting in different NO_x/HNO₃ ratios. This difference is mainly caused by a different wet deposition of HNO₃ and does not affect the NO_x concentration in the model very much [Velders and Granier, 2001]. With respect to the column, the middle and upper tropospheres contribute more to the HNO₃ tropospheric column than in the case of NO₂, but the lower troposphere and boundary layer dominate the column for both species over industrialized areas. The large NO_y column and uncertainty in it shows that the modeled NO₂ columns could be substantially affected by errors in this partitioning (see section 6).

5.2. Contribution of Different Layers

In section 5.1 and in section 4 we have considered the whole tropospheric column of NO₂. Now we will consider the contributions of different layers to the total tropospheric column, focusing in particular to how sensitive the total tropospheric column is to the boundary layer. In Plate 8 the NO₂ column is shown integrated up to the tropopause but starting at layer 1 (surface), layer 2 (bottom at 0.17 km), layer 3 (0.45 km), and layer 4 (0.94 km) as calculated by the MOZART model. From these plots it can be seen that over the more polluted continental areas the first three layers individually contribute more to the total tropospheric NO₂ column than all other layers combined. Over the industrialized areas in the eastern United States and western Europe layer 1 makes up 25-40% of the total tropospheric NO₂ column, layer 2 another 25-40%, layer 3 ~15-25%, and the other layers combined yield 10-20%. Over the remote continental areas of South America and Africa the first layer contributes 15-25%, the second layer 10-20%, the third 5-15%, while the other layers contribute 40-70%. Over the remote oceans, i.e., away from the continents, the layers above 1 km altitude contribute more than 97% to the total tropospheric column. Closer to the continents and in the North Atlantic the layers above 1 km contribute 80-90% to the total column.

The lowest few layers, representing the boundary layer, in the MOZART model clearly contribute most to the tropospheric NO₂ column (the same holds for IMAGES). The boundary layer is difficult to model properly in a global atmospheric model, since the chemical and dynamical processes in the boundary layer occur on spatial and temporal scales that are too small for the relatively coarse spatial resolution and time step of global models. Therefore, on a local scale of tens of kilometers much higher NO₂ concentrations might be found in the boundary layer than calculated by global models, which can increase the tropospheric NO₂ column significantly. As has been shown in section 2.3, GOME's sensitivity to NO₂ in the boundary layer is a complex function of albedo and wavelength, resulting in a large uncertainty in the derived tropospheric columns.

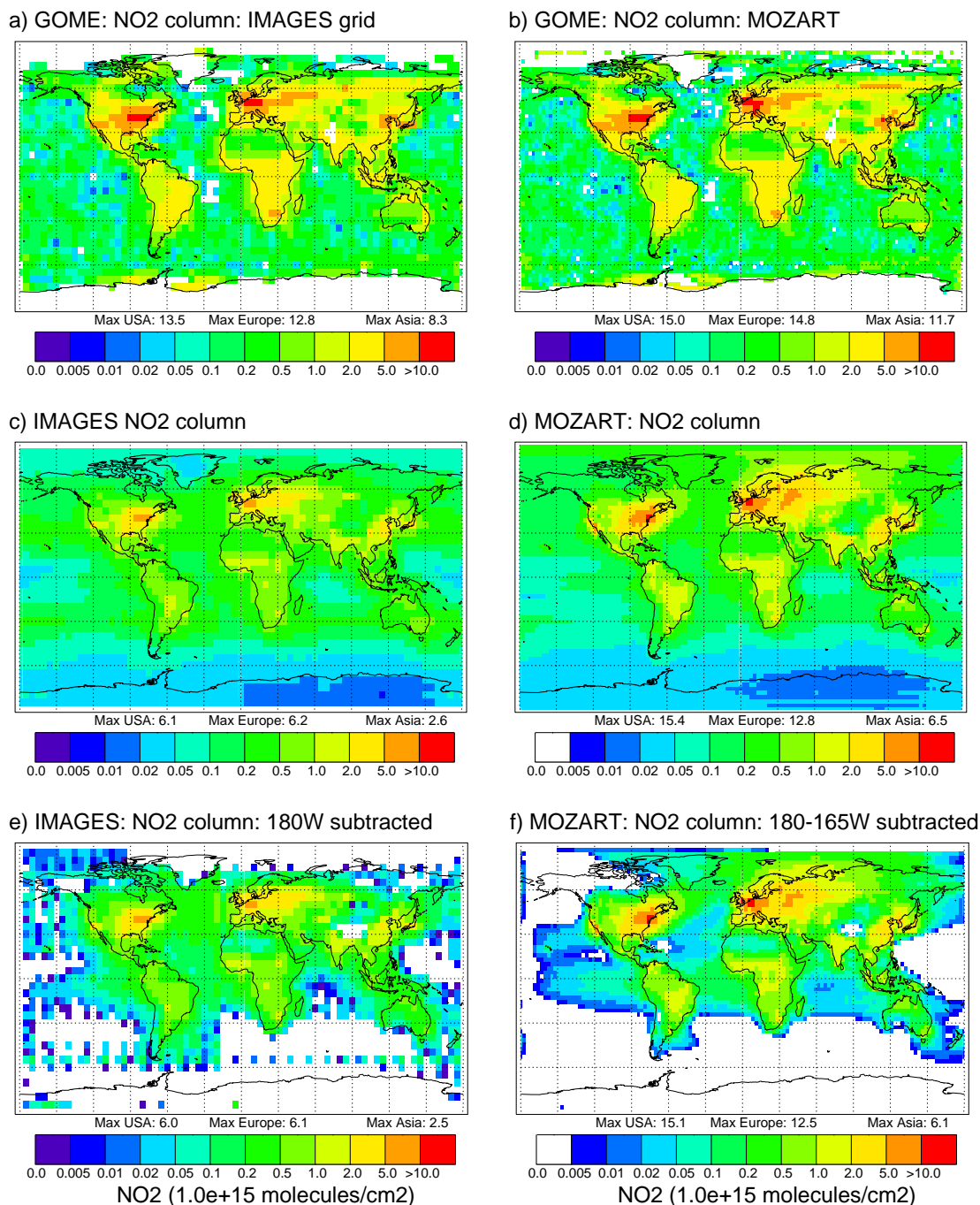


Plate 5. Yearly averaged tropospheric NO₂ columns of GOME (a, b), IMAGES (c, e), and MOZART (d, f): all plotted on (left) the IMAGES grid of 5° x 5° and (right) the MOZART grid of 2.8° x 2.8°. (c) and (d) the tropospheric column is calculated by a summation from the surface to the tropopause, while in (e) and (f) the tropospheric column is calculated by subtracting a longitude band over the Pacific from all longitudes. The latter method is similar to what is used to derive the tropospheric column from the total column in the GOME data. The columns correspond to ~1030 LT. Note a nonlinear color scale is used.

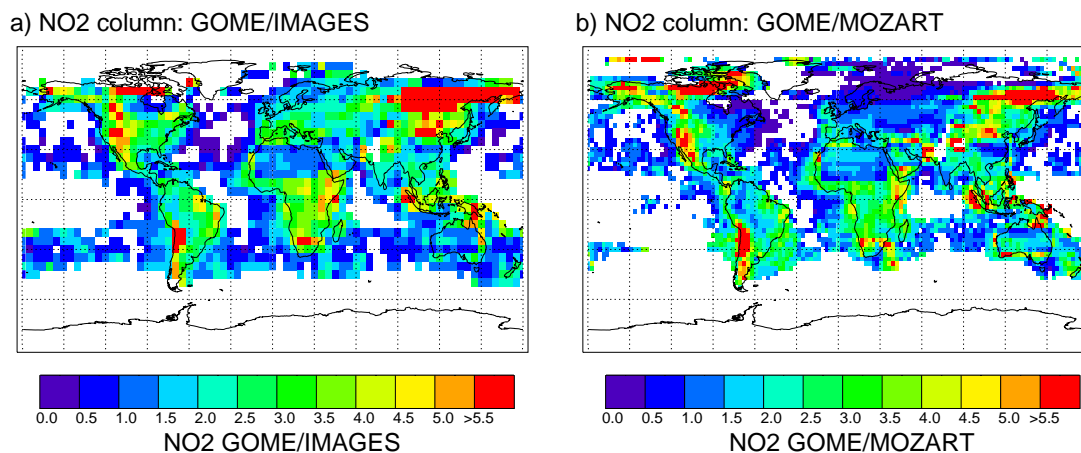


Plate 6. Ratios of the NO₂ columns from GOME relative to the IMAGES and MOZART columns, i.e., (a) GOME/IMAGES and (b) GOME/MOZART. Shown are (as a fraction) the yearly averaged columns corresponding with 1030 LT. The ratio is only plotted if both the GOME and model derived NO₂ columns are larger than 0.1×10^{15} molecules cm⁻².

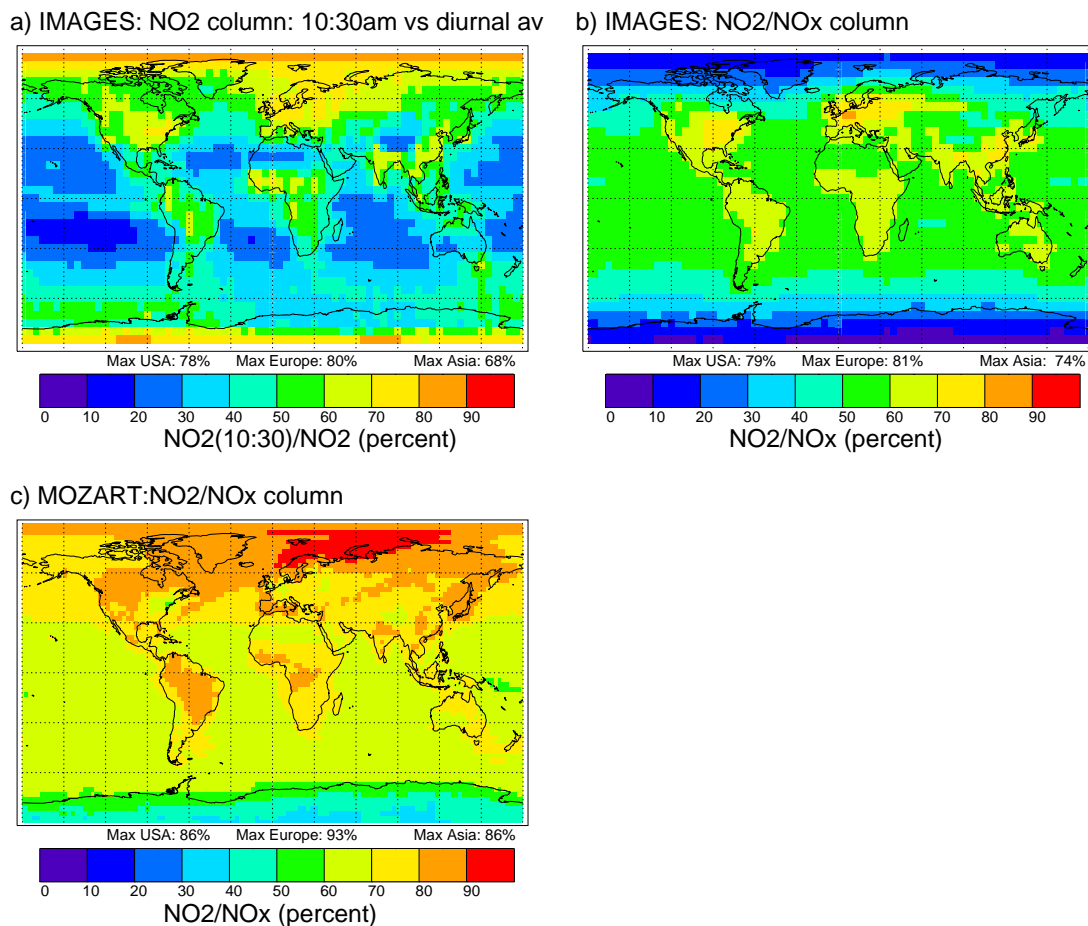


Plate 7. Ratios between (a) the tropospheric NO₂ column at 1030 LT and the diurnal average column (IMAGES), and between the diurnal average tropospheric NO₂ column and the NO_x column calculated by (b) IMAGES and (c) MOZART. Yearly averaged columns are shown.

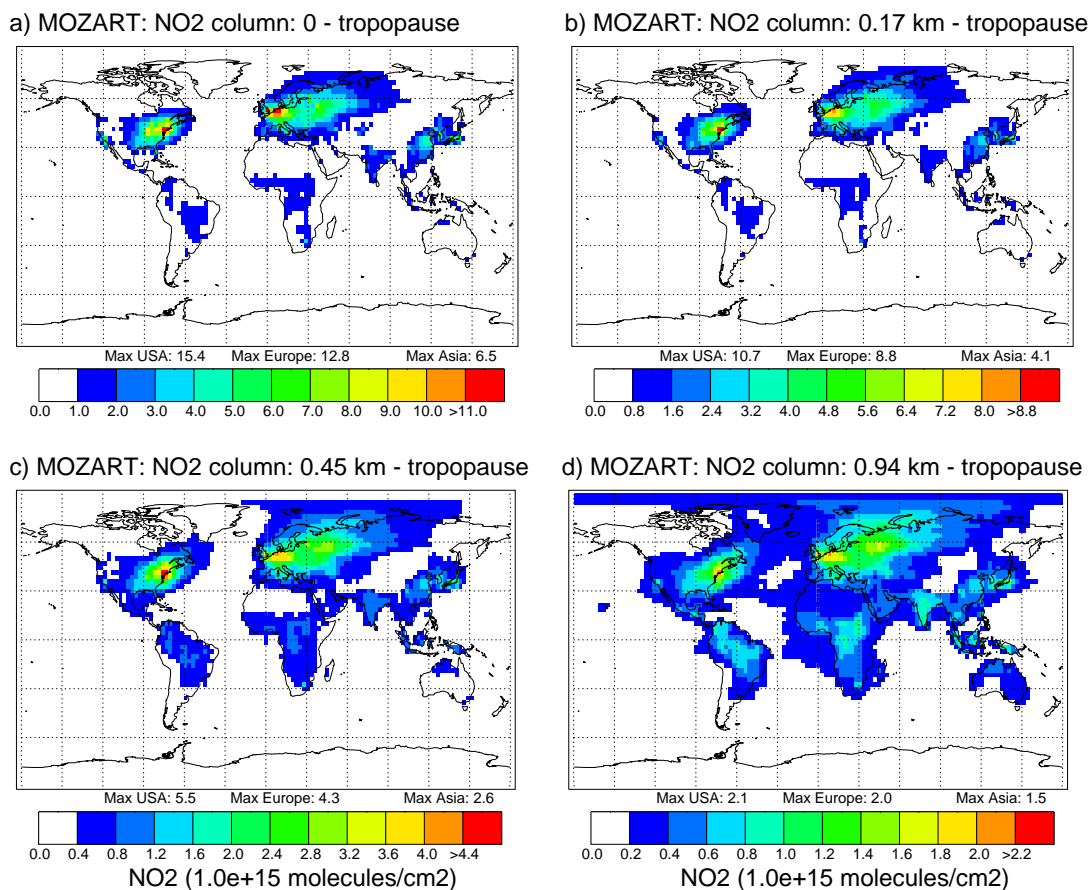


Plate 8. Different vertical integrations for calculating the yearly averaged tropospheric NO₂ column as calculated by MOZART (1030 LT). The integrations are performed up to the tropopause but starting at, (a) the surface, (b) 0.17 km, (c) 0.45 km, and (d) 0.94 km.

Radiative forcing tropospheric NO₂

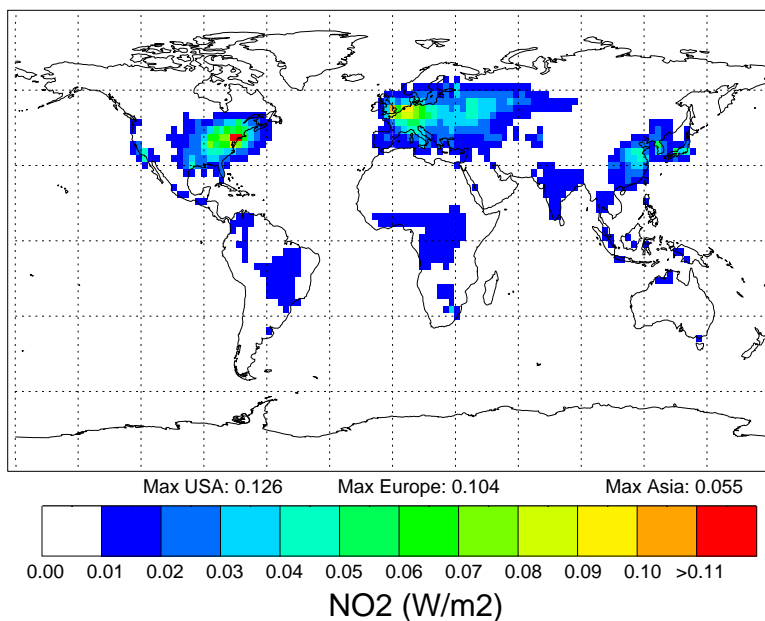


Plate 9. Annual average radiative forcing (W m⁻²) of tropospheric NO₂ calculated with MOZART. The radiative forcings are diurnal averages for cloud free conditions, calculated using the daytime average tropospheric NO₂ columns.

5.3. Heterogeneous Conversion of HNO₃ on Soot

Lary *et al.* [1997] proposed that a heterogeneous conversion of HNO₃ to NO might take place on soot. This reaction can increase the NO and thereby NO₂ concentrations in the troposphere. It has its largest affect over and downwind of polluted areas where large concentrations of both HNO₃ and soot occur, i.e., the east of China, eastern Europe, and Siberia, with smaller effects in the eastern United States and during the biomass burning season in the tropics over South America and Africa and downwind from there. Hauglustaine *et al.* [1996] showed that this reaction could reduce the HNO₃/NO_x ratio in the free troposphere at Hawaii, rendering modeled values in better agreement with measurements.

The effect of this reaction on the tropospheric NO₂ column was studied using IMAGES. Since soot is not explicitly modeled by IMAGES, it was crudely represented as sulfate (SO₄) in the calculations. This is intended only to test how sensitive the NO₂ columns are to a possible heterogeneous conversion of HNO₃. Soot and sulfate have similar sources and distributions and the reaction rate is adjusted to account for the difference in absolute values between soot and sulfate. A reaction rate of $0.3 \times 10^{-15} \text{ cm}^{-3} \text{ s}^{-1}$ was used, which is equivalent to a conversion lifetime of 2 days in the Northern Hemisphere, 0-5 km altitude. This results in tropospheric NO₂ columns in the eastern United States and western Europe of $\sim 10 \times 10^{15} \text{ molecules cm}^{-2}$, an increase of 20-30%. Larger increases occur over western Russia (80-120%), Siberia (more than 80%), and eastern China (60-80%). This heterogeneous conversion of HNO₃ has little effect in the Southern Hemisphere. Thus reactions on soot may significantly increase the modeled NO₂ column amounts.

6. Validation of Tropospheric NO₂

In the previous sections we have compared the tropospheric NO₂ columns from GOME with column amounts from chemistry transport models and found differences in the absolute values of the columns. In this section the focus is on the validation of the GOME NO₂ values. Obviously, the most straightforward way to validate NO₂ columns from GOME is to compare them with column measurements made from the ground. A large number of NO₂ column measurements from ground based stations is available but they mostly focus on stratospheric NO₂ and are in remote locations or on mountain tops (e.g., data from the Network for Detection of Stratospheric Change, NDSC). The focus in this study is on NO₂ in the polluted areas of United States, Europe, and Asia with only a few (or none at all) systematic NO₂ column measurements at these locations. We therefore compare modeled NO₂ mixing ratios with measurements in the boundary layer to assess indirectly the quality of the modeled NO₂ columns. Using the comparisons between GOME and the model calculations in the sections 4 and 5 we can then get an idea of the quality of the GOME measurements. We also compare tropospheric NO₂ columns from GOME and the models with columns inferred from mixing ratio measurements in the boundary layer.

Local measurements of NO₂ in the boundary layer are performed by many institutes all over the world. They generally take place in cities for local air quality studies. The EPA [1998] performs measurements of NO₂ at various surface locations in the United States on a daily basis. They report

yearly average NO₂ levels at urban sites of ~ 24 ppbv, suburban sites of ~ 20 ppbv and at rural sites of ~ 8 ppbv for the 1990s. Their chemiluminescence measuring technique is sensitive not only to NO₂ but also to other nitrogen-containing compounds such as PAN. This can cause an overestimation of the reported NO₂ levels, especially in rural areas [EPA, 1998]. These data are therefore less useful for our study, but they do show that very high mixing ratios are found in polluted areas, in agreement with very large local NO₂ columns over industrialized areas. Measurements in the free troposphere [e.g., Emmons *et al.*, 2000] are also less relevant for this study since the free troposphere contributes only a small amount to the tropospheric NO₂ columns.

Measurements of reactive nitrogen at rural ground sites in eastern North America were reported by Parrish *et al.* [1993]. In late summer of 1988 they measured several tropospheric trace species, including NO, NO₂, and NO_y at seven surface locations representative for rural areas in the populated regions of eastern North America. Mixing ratios over rural areas in industrialized regions of the United States and Europe are best suited for validating the GOME and modeled data, considering the grid sizes of GOME and of the models used here. Parrish *et al.* [1993] found that the daytime (1000-1800) median NO_y levels for a fully convective well mixed boundary layer were 2 to 5 ppbv. The NO_y levels vary at each site by an order of magnitude or more. The daytime NO_x/NO_y ratio was 30-50%. The observed daytime NO/NO_y ratio was $\sim 5\%$. From this we derive NO₂ daytime values ranging from 0.5 to 2.2 ppbv. The diurnal average NO₂ values are ~ 1.1 -2.8 ppbv and ~ 1.3 -3.2 ppbv at the GOME overpass time of 1030 LT.

In Figure 2 the modeled diurnal average NO₂ mixing ratios in the boundary layer are shown for the United States in August. These mixing ratios broadly agree with other model calculations [Levy *et al.*, 1999; Horowitz and Jacob, 1999] which show boundary layer values above 1 ppbv over the industrialized areas. The NO₂ measurements [Parrish *et al.*, 1993] are representative for a well-mixed boundary layer. The lowest layers in the model calculations do not show a uniform mixing ratio but have high values close to the surface and decreasing towards the top of the boundary layer. For the comparison the modeled mixing ratios are averaged over the lowest eight model layers for IMAGES (0-1.4 km) and the lowest four model layers for MOZART (0-1.6 km). The calculated average values for the eastern United States are 0.7-1.0 ppbv for IMAGES and 1-3 ppbv for MOZART. The latter values agree well with the measured values (0.5-2.2 ppbv), but those of IMAGES are too low. In the lowest layer in the models, NO₂ values are as high as 4 ppbv (IMAGES) and 15 ppbv (MOZART). In winter, with a smaller boundary layer, values more than twice as large are found both for the average in the boundary layer and in the lowest layer.

A mixing ratio of 1 ppbv in a well mixed boundary layer of 1 km contributes $2.69 \times 10^{15} \text{ molecules cm}^{-2}$ to the total column. The observed 1.3-3.2 ppbv of NO₂ at ~ 1030 LT therefore yields ~ 3.5 - $8.6 \times 10^{15} \text{ molecules cm}^{-2}$ for a boundary layer of 1 km. Adding to this an NO₂ column in the free troposphere of $\sim 1.5 \times 10^{15} \text{ molecules cm}^{-2}$ (August data from MOZART model), we obtain a total tropospheric column of 5.0 - $10.1 \times 10^{15} \text{ molecules cm}^{-2}$. The NO₂ columns retrieved from GOME in the eastern United States in August are ~ 7 - $10 \times 10^{15} \text{ molecules cm}^{-2}$, with maximum values up to $16 \times 10^{15} \text{ molecules cm}^{-2}$ in very localized areas. This is within the

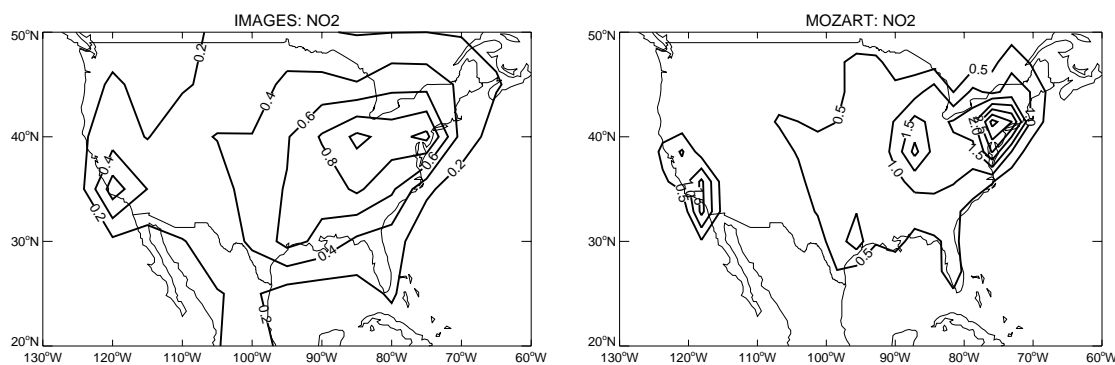


Figure 2. Diurnal average NO₂ mixing ratios (ppbv) in the boundary layer for August calculated with (a) IMAGES and (b) MOZART. The mixing ratios are averaged over the lowest 1.5 km of the modeled atmosphere.

range of the columns inferred from the measured NO₂ mixing ratios and slightly larger than the columns calculated with the MOZART model ($4\text{--}8 \times 10^{15}$ molecules cm⁻² at ~ 1030 LT in August, with a local maximum of 13×10^{15} molecules cm⁻²). The values from IMAGES are $2.5\text{--}4.2 \times 10^{15}$ molecules cm⁻².

The following qualifications concerning this comparison need to be taken into account:

1. The boundary layer thickness is not known exactly, affecting the comparison of the modeled columns with the surface mixing ratio data. The thickness of 1 km we assumed here is probably on the low end, resulting in a low estimate of the column inferred from the surface mixing ratios.

2. The boundary layer in the model calculations does not have a uniform mixing ratio but exhibits high surface values decreasing toward the top of the boundary layer. For this comparison, the average of the mixing ratios in the lowest 1.5 km has been used. This is not necessarily the same as a well-mixed boundary layer because of possible nonlinear NO_x chemistry.

3. The relatively large grid size of the models affects the NO_x chemistry since the NO emissions are emitted into these boxes and diluted more than in reality, resulting in lower local concentrations and a different conversion to NO₂.

4. High NO₂ mixing ratios and columns occur on a local scale in urban and suburban areas (see large mixing ratio in urban and suburban areas as measured by the EPA [1998]), which cannot be modeled in a global model and probably difficult to observe by GOME, but can affect the NO₂ columns from both.

5. The data (GOME, model, and surface NO₂ mixing ratios) all show considerable spatial variability, which complicates the comparison.

These considerations do not change the overall picture that the tropospheric NO₂ columns from GOME over eastern North America are of the same order as those derived from both the observed in situ NO₂ mixing ratios and those from the model calculations with MOZART.

7. Radiative Forcing of Tropospheric NO₂

In addition to playing a role in the formation of ozone in the troposphere, NO₂ also absorbs solar radiation and thereby may contribute to the radiative forcing of climate. Solomon *et al.* [1999] showed that over Boulder (Colorado) NO₂ can

cause a local instantaneous absorption of radiation in the troposphere of up to $\sim 2\text{--}10$ W m⁻², likely resulting from air pollution. Over polluted areas in the eastern United States, northwestern Europe, and the east of China a larger local radiative absorption can be expected. Absorption of solar radiation in the troposphere by NO₂ of up to 30 W m⁻² has been estimated to take place in clouds during thunderstorms [Solomon *et al.*, 1999]. Note that the local radiative absorption values given by Solomon *et al.* [1999] correspond to the net absorption occurring in the tropospheric air column. This is different from the radiative climate forcing (i.e., the change in flux at the tropopause) since it includes absorption which would have occurred at the ground anyway. The difference is largest for clear skies and cloudy skies where the absorber is located below the cloud (for optically thick clouds where the absorber is located in or above the cloud the difference is relatively small). In this section we use the modeled NO₂ columns from IMAGES and MOZART to make a global map of the radiative forcing by NO₂ in the troposphere for clear sky conditions.

A line-by-line radiative transfer calculation which incorporates the DISORT [Stamnes *et al.*, 1988] scattering code has been used in this study. This is the same basic model as used in Solomon *et al.* [1999]. A perturbation profile of NO₂ with all of the NO₂ in the lowest 1 km is assumed since the models show that most of the NO₂ is in the boundary layer in polluted regions. Overlap with O₃ and O₄ is included although their influence is small except for O₃ absorption in the UV. A ground albedo of 0.05 is assumed, which is reasonable for continental urban areas in the blue region of the spectrum. An aerosol layer is included in the boundary layer with an optical depth of 0.1, a single scattering albedo 0.85, and an asymmetry factor of 0.7, which are very conservative estimates for polluted regions. The diurnal cycle is integrated over for the 15th day of each month as a function of latitude for clear sky conditions. This yields a local radiative forcing efficiency per NO₂ molecule (W m⁻²/cm⁻²) which can be multiplied by the NO₂ column to obtain the local radiative forcing.

Plate 9 shows the diurnal annual average radiative forcing of tropospheric NO₂ as calculated using the daytime averaged tropospheric NO₂ columns from MOZART. Since the daytime NO₂ columns were not directly available from the MOZART calculations, we used the results from the IMAGES model to calculate the ratio between the daytime average tropospheric

NO₂ columns and the diurnal average columns. This ratio was then used on a monthly basis to calculate the daytime average columns for the MOZART calculations. The maximum monthly values attained by the local radiative forcing of tropospheric NO₂ are 0.1-0.15 W m⁻². Much larger values can be expected in cities where very high NO₂ concentrations are observed; events that are not well represented in the global model calculation presented here. These local "hot spots" over industrialized areas could be important for local heating in urban areas. On a more continental scale a radiative forcing of 0.04-0.1 W m⁻² is calculated over the eastern United States, western Europe, and eastern Russia. The calculation of radiative forcing illustrates the regional distribution and magnitude, at least to a factor of 2 accuracy. The calculated values are of the same order as the radiative forcing of the well-mixed greenhouse gas N₂O (0.14 W m⁻²) but smaller than the radiative forcing of the halocarbons (0.25 W m⁻²) and of tropospheric ozone (0.2-0.6 W m⁻²) [Intergovernmental Panel on Climate Change, 1996; Kiehl et al., 1999; World Meteorological Organization, 1999]. Globally averaged, the estimated radiative forcing of tropospheric NO₂ is relatively small having a value of ~0.005 W m⁻². The radiative forcing of NO₂ shows a small seasonal cycle, with ~15% larger values than average in winter/spring and ~15% lower values in autumn. The seasonal cycle is not very large because the larger forcing in the summer compared with the winter, due to the zenith angle dependence, is compensated by the smaller NO₂ columns. Assuming a larger ground albedo of 0.15 in the calculation increases the estimated radiative forcing by a factor of 2. Forcing maps derived from the IMAGES calculations are about half those derived from MOZART.

The radiative forcing of tropospheric NO₂ as estimated here is based on cloud free conditions. Since the bulk of the NO₂ resides in the boundary layer below the clouds, clouds effectively shield the solar radiation from the NO₂. Even clouds with relatively small optical depths will reduce the globally averaged radiative forcing. For example, a cloud with an optical depth of 10 will reduce the radiative forcing by 65%. Thus, as a lower limit we can assume that clouds will reduce the estimated local radiative forcing by nearly the fraction of local cloud cover. Assuming an average cloud cover of 50% the maxima in the local radiative forcing over the industrialized areas becomes ~0.05-0.08 W m⁻².

8. Conclusions

Tropospheric NO₂ columns derived from GOME data have been compared with that from model calculations by two global three-dimensional chemistry transport models, IMAGES and MOZART. The global patterns in NO₂ columns are well reproduced by the models: large tropospheric columns over the industrialized areas of the eastern United States, Europe, and Asia, as well as over cities in Africa and South America; increased NO₂ columns are found over the biomass burning areas of Africa and South America. The variation through the seasons is also similar in both the GOME and the model data sets, having larger values in winter and smaller ones in summer. The quantitative comparison is limited by the uncertainty in the tropospheric NO₂ columns derived from GOME caused by the uncertainty in the air mass factors and cloud cover and due to the difficulty of modeling the dynamics of the boundary layer properly in global models. A tentative comparison of the absolute values shows that the

measured tropospheric column amounts inferred from GOME are of the same order as those calculated by MOZART over industrialized areas. The maximum values in MOZART are similar to those of GOME for the United States (a factor of 0.8-1.5 smaller) and Europe (0.7-2), but a factor of 1.5-3.5 smaller for Asia (on the same grid). If averaged over a larger area (subcontinental) the high maximum values in very localized areas in MOZART and in GOME are somewhat reduced. The agreement between the model calculations and the GOME measurements is somewhat worse for South America and Africa, areas with lower NO₂ columns in which biomass burning, savannah fires, and soil emissions are the dominant NO_x sources. The maximum values of IMAGES are significantly smaller than those of GOME: about a factor of 2-3 for the United States, ~1.5-4 for Europe, and ~3-5 for Asia, using the same grid for both.

Over polluted areas the largest contribution to the tropospheric NO₂ column originates from the boundary layer. The boundary layer is difficult to model properly in a global atmospheric model, because the chemical and dynamical processes in the boundary layer occur on spatial and temporal scales that are too small for the relatively coarse spatial resolution and time step of global models. Therefore much higher NO₂ concentrations might be found on smaller scales (several kilometers) in the boundary layer than those calculated by global models, which can locally increase the tropospheric NO₂ column significantly.

The effect of a possible heterogeneous conversion (on soot) of HNO₃ to NO on the modeled NO₂ columns in the troposphere has been studied with the IMAGES model. We found that this reaction can increase the modeled tropospheric NO₂ columns with 20-30% in the eastern United States and western Europe and around 100% in western Russia, Siberia, and eastern of China. The modeled ratios NO₂/NO_x and NO_x/NO_y have been analyzed and found to be close to observations over eastern North America, adding confidence to model calculations of the NO₂ columns.

The modeled tropospheric NO₂ columns from MOZART as well as the columns measured by GOME agree with NO₂ columns derived from observed NO₂ mixing ratios in the boundary layer in eastern North America: the NO₂ columns from GOME are 7-10 × 10¹⁵ molecules cm⁻², from MOZART are 4-8 × 10¹⁵ molecules cm⁻², and those derived from the surface mixing ratios are 5-10 × 10¹⁵ molecules cm⁻². In IMAGES, NO₂ is transported out of the boundary layer too fast resulting in lower NO₂ columns (2.5-4 × 10¹⁵ molecules cm⁻²). Qualitatively, the pattern of tropospheric NO₂ columns as measured by GOME agree well with the ones calculated by MOZART and IMAGES. A quantitative comparison is more difficult, but indicates that the GOME NO₂ columns, those calculated by MOZART, and those derived surface mixing ratios are all of the same order for eastern North America.

An estimate of the radiative forcing of NO₂ has been made using the calculated tropospheric NO₂ columns from MOZART. The local maxima over the eastern United States and western Europe in the radiative forcing of tropospheric NO₂ is 0.1-0.15 W m⁻², while values of 0.04-0.1 W m⁻² are estimated on a continental scale (cloud free conditions). The all sky estimates are about a factor of 2 smaller than the clear sky values. These values are of the same order as the radiative forcing of N₂O but smaller than that of the halocarbons and tropospheric ozone. The globally averaged radiative forcing is negligible, ~0.005 W m⁻².

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